CLAIMS

The embodiment of the invention in which an exclusive property or privilege is claimed is defined as follows:

- 1. An improved electrolyte for the electrolysis of alumina (Al₂O₃), the electrolyte comprising a mixture of aluminum fluoride (AlF₃) and potassium fluoride (KF).
- 2. The electrolyte recited in claim 1 wherein the aluminum fluoride (AlF₃) is present in a molar ratio with potassium fluoride (KF) greater than 1:1.
- 1 3. The electrolyte recited in claim 1, the electrolyte further comprising from about 2 to 6 wt. % of aluminum oxide (alumina/Al₂O₃).
 - 4. The electrolyte as recited in claim 1 wherein the concentrations of the electrolyte's components remain constant during electrolysis.

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- 5. The electrolyte as recited in claim 1 where the electrolyte is used to electrolyze the alumina to aluminum at from about 660°C to 1000°C.
- 1 6. The electrolyte as recited in claim 3 wherein the electrolyte remains liquid during electrolysis.
 - 7. The electrolyte as recited in claim 6 wherein inert anodes and wettable cathodes can be used with the electrolyte.

8. A method for using inert anodes and wettable cathodes in the electrolytic production of aluminum comprising using the anodes and cathodes in an electrolysis process at a temperature from between 660°C and 1000°C.

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- 9. The method as recited in claim 8 wherein the anode is comprised of a material selected from the group consisting of metals, metal alloys, metal oxides, and cermets.
- 10. The method as recited in claim 8 wherein the alloys are copper alloys (aluminum-bronze) and are from about 1 wt.% to 20 wt.% aluminum (Al).
- 11. The method as recited in claim 8 wherein the cathode is made of a material selected from the group consisting of metal borides, metal boride-carbon composites, metal boride-containing coatings on substrates, graphite, molybdenum, and tungsten.
- 12. The method as recited in claim 8 where the anode contains an oxygen-containing film.
 - 13. The method as recited in claim 12 wherein the oxygen containing film is formed during the electrolysis process.
- 14. The method as recited in claim 11 wherein the material coats one side of the inert anode to create a bipolar electrode for use in multipolar electrolytic cells.
- 15. The method as recited in claim 8 wherein the distance between the anode and cathode remains constant.
- 1 16. The method as recited in claim 8 wherein the electrodes can be arranged in a vertical configuration.

1	17.	The method as recited in claim 8 wherein the electrodes can be arranged
2	in a slanted o	configuration.
1	18.	The method as recited in claim 8 wherein O ₂ is the only gas generated by
2	electrolysis.	
1	19.	The method as recited in claim 1.6 wherein the vertical configuration of the
2	cell creates	electrolyte recirculation pathways.
1	20.	A method for electrolyzing alumina below 1000°C, the method
2	comprising:	
3		a) supplying an electrolyte containing more than 35 mol% potassium
4	fluoride (KF)	and more than 30 mol% aluminum fluoride (AIF ₃);
5		b) injecting alumina with the electrolyte; and

subjecting the electrolyte to a voltage.

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